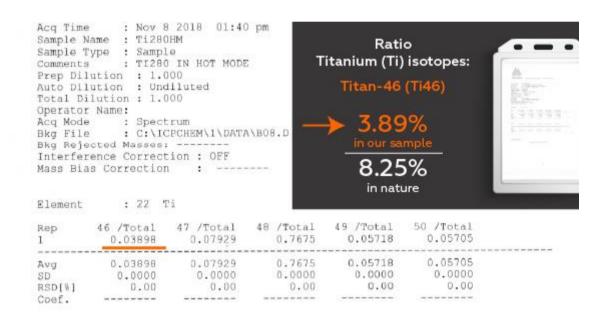


## Proof of achievements in research of "Synthestech"

## Isotopic analysis and certain results of our research



We are grateful to everyone who participates in investing our project. This confidence is of great importance for us, and we will justify it.

At the same time, sometimes questions on the evidence that the obtained elements are acquired as a result of Cold Nuclear Transmutation processes of chemical elements implementation. Indeed, this is an important question.

This is a very difficult task. Moreover, artificially synthesized elements are impossible to distinguish from the natural usual methods of analysis. This is the same chemical element. (It is right, their biological activity, as we believe, is different, but this is a

completely new sphere, where we operate). However, we successfully coped with this task and we have indisputable proof of our success.

Especially for this purpose we have equipped analytical laboratory with advanced methods of analysis, which allows us to present evidence of our results.

All chemical elements have from one to several isotopes.

Isotopes are types of atoms of one chemical element differing in mass of nuclei. Isotope elements have the same number of protons but different number of neutrons in the nuclei. Natural balance of stable isotopes of one chemical element has a completely strict value. For example, sulfur (S) isotopes in nature have the following ratio - 32S (95,02%), 33S (0,75 %), 34S (4,21 %), 36S (0,02 %). Deviations are extremely rare, within the percentage. This is true even for the sulfur, that is found in meteorites. That is, all processes in the foreseeable space lead to a constant natural ratio of isotopes. And so touches all chemical elements. Today, a person can change this ratio, but only with the help of enrichment facilities worth hundreds of millions of dollars. Such facilities are used in the nuclear industry and access to them is closed. In addition, it is necessary to develop an enrichment technology for each isotope, which also costs many millions of dollars.

At the same time, during the Cold transmutation of nuclei processes of, newly obtained elements, as a rule, also reproduce the natural ratios of isotopes. However, as a rule, it does not always mean - under some processes that we use to carry out Cold Transmutation, we change isotopic ratio of individual elements.

This concerns, for example, Titan (Ti) - the natural ratio of its isotopes is as follows: **46Ti** (**8,25** %), 47Ti (7,44 %), 48Ti (73,72 %), 49Ti (5,41 %), 50Ti (5,18 %).

In our samples, which mass spectrometry analysis card is applied, the ratio differs significantly from the natural one - **46Ti** (**3,89%**), 47Ti (7,29%), 48Ti (76,75%), 49Ti (5,71%), 50Ti (5,75%). As it is evident, the **46Ti** isotope is 2 (!!) times less in our sample than the natural one.

## Such Titan (Ti) is not available in nature, it is impossible to buy or somehow obtain it.

Nobody in the world can make this. This fact is a reliable proof of the implementation of Cold Transmutation within the framework of *«Synthestech»* project.

Everyone can make sure in it. We plan to send such samples to non-believer - to check them.

In the nearest future, we will move to our new Center and the implementation of the Cold Synthesis project of valuable elements will move into a practical phase.

News of project «Synthestech»

Please, note, that project tokens can be purchased with a bonus of 47,5%. Project Token costs 1,5 USD, but today, given the provided bonus, you can buy it for a bit more than 1 USD. This is a temporary offer. So do not miss your chance!

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